Chemical Processes

Things you should learn:

- 1. Two major methods for producing a desired molecule/compound/substance
- 2. Types of molecules: small, large, polymers
- 3. Basic formulas for describing molecules in terms of constituent atoms
- 4. The use of moles in defining chemical quantities
- 5. Reactors as a means for producing substances
- 6. Problems associated with exo- vs. endo- thermic reactions
- 7. Mass/energy balance in a closed system
- 8. How separation processes isolate substances
- 9. Properties of substances that can be used in these separation processes
- 10. What is a hydrometer and how does it work
- 11. Creation of substances using a reactor
- 12. What goes in must come out
- 13. How to use mass/energy balance as a deductive tool
- 14. What are unit operations
- 15. Distillation as a separation process—McCabe-Thiele diagrams
- 16. What is an azeotrope
- 17. Using a McCabe-Thield diagram to calculate number of separation stages

Things you should be able to do:

- 1. Carry out bilinear interpolation of temperature, concentration, density tables
- 2. Calculate the number of unit-op stages required to purify a desired substance
- 3. Design a sequence of separation processes to isolate constituents of a dry mixture
- 4. Determine specific gravity of a liquid/solid

Things you should lie awake at night thinking about:

- 1. TiO₂ is an \$80B yearly business. A 1% improvement in efficiency in producing this product is worth 800M/year.
- 2. Distillation as a separation process represents about 75% of all chemical processes (primarily due to the petroleum industry).
- 3. What are the problems associated with bio-substances, e.g., how might one separate different length strands of DNA?